

# Questions & Answers

## Evidence

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patients by asking and answering specific clinical care questions The book is organized into sections by presentation gastrointestinal bleeding for example each of which contains chapters on specific questions such as What is the best clinical risk score for low risk GIB patients Each clinical question comes with a detailed evidence based response and a summary that gives best practices recommendations and references Additionally at the end of each section is a chapter titled Expert Corner which asks the same clinical questions to a surgical or gastrointestinal specialist and includes key pearls these experts have for emergency medicine practitioners

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by a single equation of state called the ideal gas law equation this shows the relationship between a gas's pressure  $p$  temperature  $t$  volume  $v$  and amount in moles  $n$

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web each gas in a mixture of gases exerts its own pressure independently of the other gases present therefore the pressure of each gas within a mixture is called the partial pressure of the gas dalton's law of partial pressures states that the total pressure of a mixture of gases is equal to the sum of the partial pressures of the component gases

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web at very high pressures a gas will occupy a larger volume than predicted by the ideal gas law which of the following statements is true at low temperatures intermolecular forces become important and the pressure of a gas will be lower than predicted by