

Name: _____

Group: _____

Date: _____

Balancing Chemical Equations - Homework Sheet
Grade 10 Science

PART 1: Balance the following chemical equations

***Note, you may need to work out these balancing equations on extra paper**

1. $\text{N}_2 + \text{H}_2 \rightarrow \text{NH}_3$
2. $\text{S}_8 + \text{O}_2 \rightarrow \text{SO}_3$
3. $\text{HgO} \rightarrow \text{Hg} + \text{O}_2$
4. $\text{Zn} + \text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$
5. $\text{SiCl}_4 + \text{H}_2\text{O} \rightarrow \text{H}_4\text{SiO}_4 + \text{HCl}$
6. $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$
7. $\text{H}_3\text{PO}_4 \rightarrow \text{H}_4\text{P}_2\text{O}_7 + \text{H}_2\text{O}$
8. $\text{Si}_2\text{H}_3 + \text{O}_2 \rightarrow \text{SiO}_2 + \text{H}_2\text{O}$
9. $\text{Al}(\text{OH})_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + \text{H}_2\text{O}$
10. $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
11. $\text{Fe}_2(\text{SO}_4)_3 + \text{KOH} \rightarrow \text{K}_2\text{SO}_4 + \text{Fe}(\text{OH})_3$
12. $\text{FeS}_2 + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3 + \text{SO}_2$
13. $\text{Al} + \text{FeO} \rightarrow \text{Al}_2\text{O}_3 + \text{Fe}$
14. $\text{Na}_2\text{CO}_3 + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O} + \text{CO}_2$
15. $\text{K} + \text{Br}_2 \rightarrow \text{KBr}$
16. $\text{P}_4 + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$
17. $\text{C}_2\text{H}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

Questions In Standard Grade Chemistry

Princeton Review (Firm)



Questions In Standard Grade Chemistry:

Questions in Standard Grade Chemistry Sandy Herd, 2000 A key companion to Standard Grade Chemistry Revision Notes this book offers vital examination practice and represents a useful revision tool for students at General and Credit level Helpful in monitoring student progress throughout the course sets of questions are organised by topic and focus on testing ability to apply knowledge and understanding to problem solving and to carry out chemical calculations In this way with use of the pull out answer section students can easily focus on and improve areas of weakness to raise exam response Contents Chemical reactions and speeds of reactions Atoms and the periodic table Bonding formulae and properties Fuels Hydrocarbons Chemical calculations Acids and alkalis Making electricity Metals Corrosion Plastics and synthetic fibres Fertilisers Carbohydrates

Revision Questions for Standard Grade Chemistry D. A. Buchanan, J. R. Melrose, 1997

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Other chapters have been updated where relevant for instance to reflect changes to the Relative Atomic Mass data The chemical dictionary chapter has been enlarged to include more key words New exam questions have been included together with homework and whole course questions at the end of the book

Horn's Graduated standard arithmetical exercises. With answers. With answers Joseph Stephenson Horn,1872

Cracking the AP Chemistry Exam, 2018 Edition Princeton Review,2017-08 Provides subject review for the AP chemistry exam and includes two full length practice tests

Analytical Chemistry for Technicians John Kenkel,2013-08-13 Written as a training manual for chemistry based laboratory technicians this thoroughly updated fourth edition of the bestselling Analytical Chemistry for Technicians emphasizes the applied aspects rather than the theoretical ones The book begins with classical quantitative analysis and follows with a practical approach to the complex world of so

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ENC Focus ,2000

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Transactions of the American Institute of Chemical Engineers ,1924

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